

**Contribution of the dispersion to the total intermolecular interaction energy**

Molecule pair	% of the total energy of interaction
<u>Ne</u> -Ne	100
<u>CH<sub>4</sub></u> -CH <sub>4</sub>	100
<u>HCl</u> -HCl	86
<u>HBr</u> -HBr	96
<u>HI</u> -HI	99
<u>CH<sub>3</sub>Cl</u> -CH <sub>3</sub> Cl	68
<u>NH<sub>3</sub></u> -NH <sub>3</sub>	57
<u>H<sub>2</sub>O</u> -H <sub>2</sub> O	24
HCl-HI	96
H <sub>2</sub> O-CH <sub>4</sub>	87